## A Low Cost, High Capacity Regenerable Sorbent for Pre-Combustion CO<sub>2</sub> Capture

Contract No. DE-FE0000469
Project Review Meeting



Gökhan Alptekin, PhD Ambal Jayaraman, PhD Robert Copeland, PhD

> Pittsburgh, PA August 25, 2011

#### **Project Objective**

- The objective of this work is to develop a new pre-combustion CO<sub>2</sub> capture technology and demonstrate its technical and economic viability
- A low cost, high capacity regenerable sorbent removes CO<sub>2</sub> above the dew point of the synthesis gas
- The sorbent is a mesoporous carbon grafted with surface functional groups that remove CO<sub>2</sub> via physical adsorption
- Budget Year 1
  - Sorbent optimization and production scale-up
  - Bench-scale evaluations
  - Process design and optimization
- Budget Year 2
  - Demonstrate sorbent life for 10,000 cycles
  - Slipstream demonstration using actual synthesis gas
  - Based on field data and optimum design, conduct an economic analysis to estimate the cost of CO<sub>2</sub> capture

TDA Research

#### **Project Partners**



#### **TDA Research**









#### **Project Duration**

- Start Date = November 15, 2009
- End Date = September 30, 2012 (no-cost extension is being worked out)

#### **Budget**

- Project Cost = \$2,500,000
- DOE Share = \$2,000,000
- TDA and its partners = \$500,000

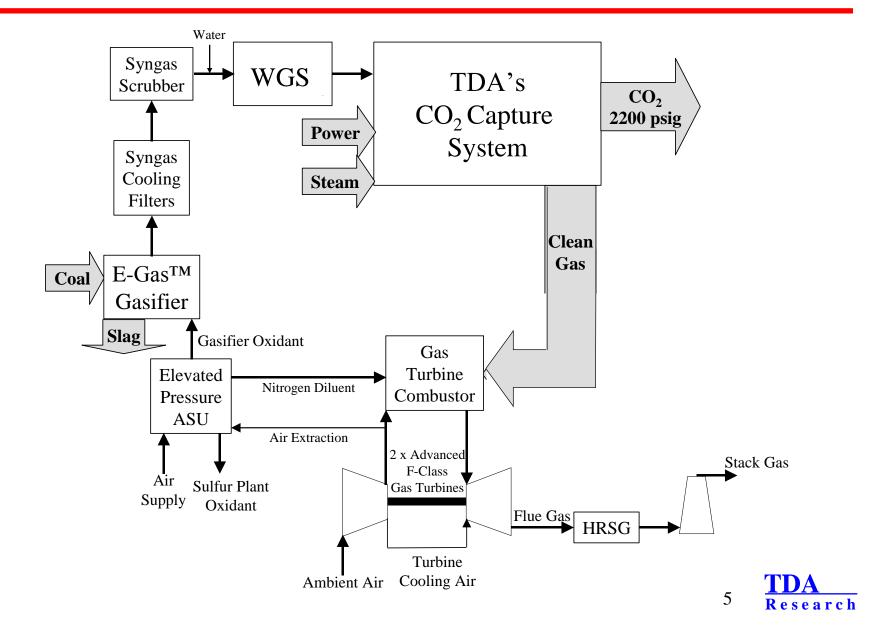


### **TDA's Approach**

- The sorbent consists of a carbon material modified with surface functional groups that remove CO<sub>2</sub> via strong physical adsorption
  - CO<sub>2</sub>-surface interaction is strong enough to allow operation at elevated temperatures
  - Because CO<sub>2</sub> is not bonded via a covalent bond, the energy input for regeneration is low
- Heat of adsorption of CO<sub>2</sub> is measured as 4.9 kcal mol per mole for TDA sorbent
  - Selexol ~4 kcal/mol
  - Amine solvents ~14.4 kcal/mol
  - Chemical absorbents 20-40 kcal/mol (Na<sub>2</sub>CO<sub>3</sub>→NaHCO<sub>3</sub> 30 kcal/mol)
- Net energy loss in sorbent regeneration is similar to Selexol
  - A much better IGCC efficiency due to higher temperature CO<sub>2</sub> capture
  - Warm gas clean-up improves cycle efficiency 2 to 4%



## **IGCC-Integrated CO<sub>2</sub> Capture System**



#### **Regeneration Options**

- Physical adsorbent provides flexibility in regeneration
  - Temperature swing
  - Pressure swing
  - Concentration swing
  - Combinations
- Isothermal operation is critical to eliminate heat/cool transitions which reduces cycle time and increases sorbent utilization
- Steam consumption can be reduced significantly if steam purge is carried out at low pressure

Syngas Inlet 236°C, 500 psia 40% CO<sub>2</sub> P<sub>CO2</sub> = 200 psia

Steam/CO<sub>2</sub>
235°C, 145 psia
86% CO<sub>2</sub>
P<sub>CO2</sub> = 125 psia

Regeneration

**Adsorption** 

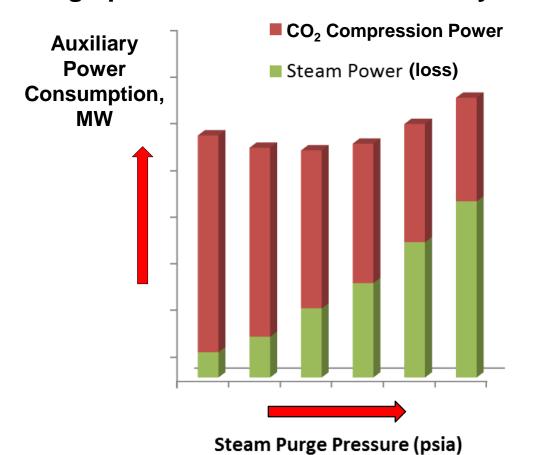
Syngas Inlet 244°C, 492 psia <1% CO<sub>2</sub>

 $P_{CO2} = 5 psia$ 

Steam
245°C, 150 psia
0% CO<sub>2</sub>
P<sub>CO2</sub> = 0 psia

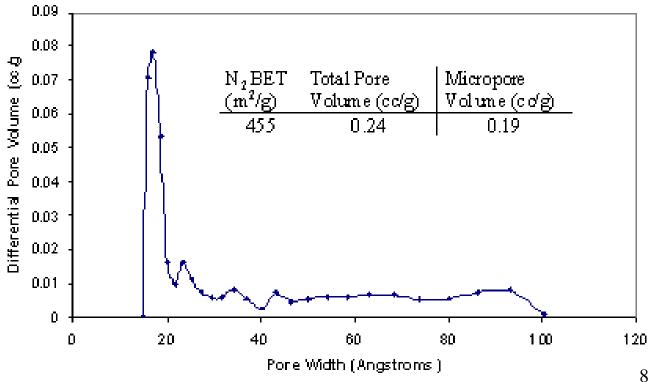
# Trade-off – Regeneration Pressure vs. Steam Consumption

Higher regeneration pressure reduces power input for CO<sub>2</sub> compression, while pure concentration swing requires large amounts of high pressure steam from steam cycle



#### **TDA's Sorbent**

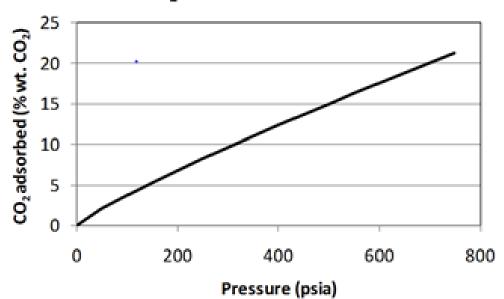
- Mesoporous carbon has been developed for ultra-capacitors
  - Meso-range pores (20 to 100 Å) are large enough to allow transport of liquid electrolyte in and out of the pores
  - Macro-porosity is avoided to achieve high surface area
  - Surface is modified with functional groups to enhance CO<sub>2</sub> selectivity





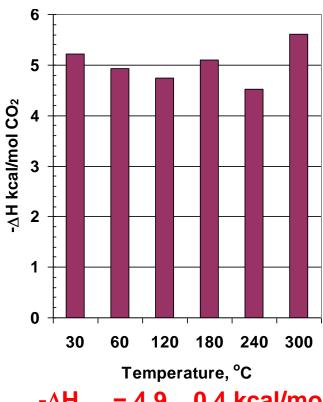
### CO<sub>2</sub> Isotherm and Heat of Adsorption

#### CO<sub>2</sub> isotherm at 240°C



Langmuir Coefficient (q <sub>s</sub> )	386.4	mmol/g		
Langmuir Coefficient (B)	4.15E-04	1/atm		
Langmuir Coefficient (n)	0.869			
Diffusion Coefficient (D/R²)	1.32E-03	1/s		
Reference Temperature for B	240	°C		
Heat of Adsorption (∆H)	4.8	kcal/mol		

#### **Calorimetry Measurements**



 $-\Delta H_{ads} = 4.9$  0.4 kcal/mol

Isosteric heat of adsorption calculations and DSC experiments confirm the low heat of adsorption



#### **Sorbent Production Scale-up**







- Early samples are prepared using a batch process
  - 11" diameter
  - Computer controlled
  - 1000 C temp. limit
  - ~5 kg carbon/run

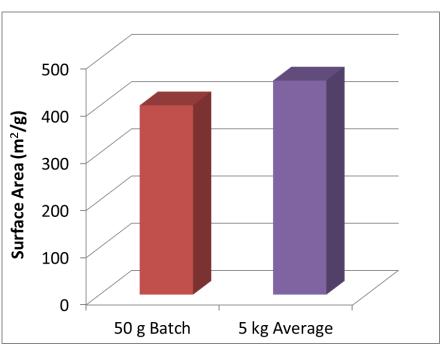
 60 kg sorbent is prepared for field demonstrations

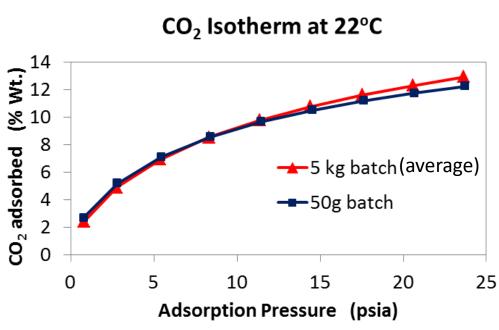


#### **Sorbent Production Scale-up**

#### **Surface Area**

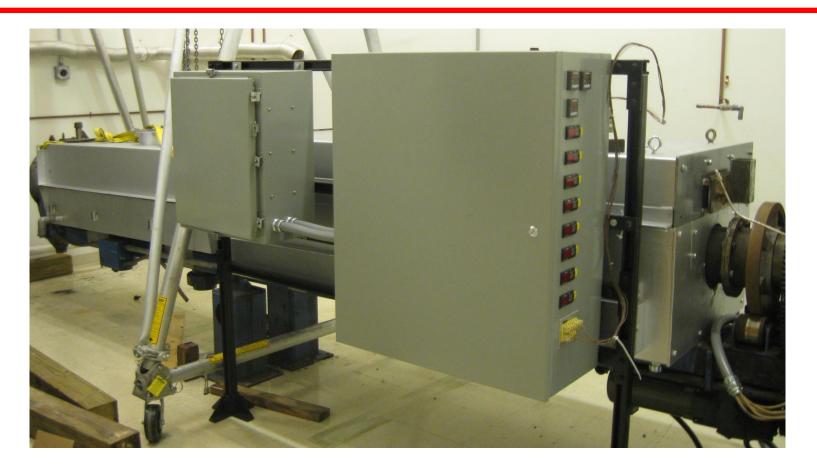






- The scaled-up sorbent showed surface area and CO<sub>2</sub> capacity similar to the sorbent produced at small batch size
  - Low temperature isotherms measurements were used for convenience

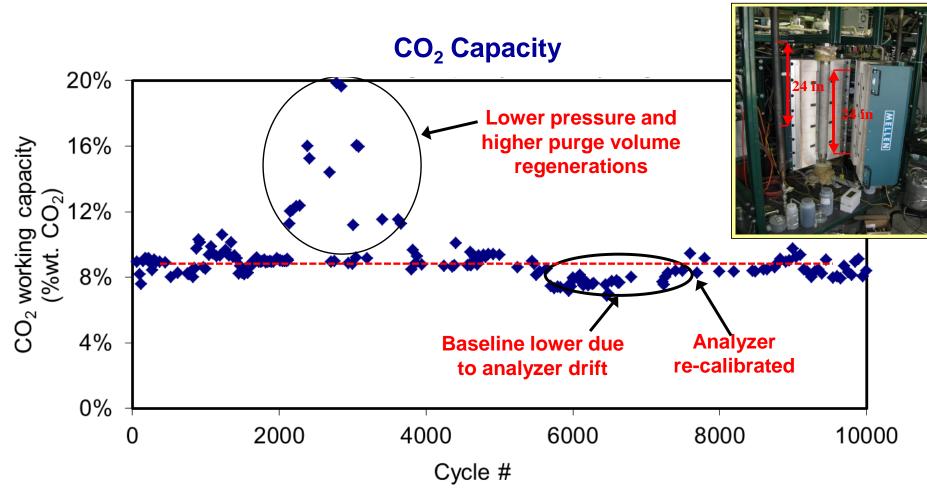
#### **Sorbent Production Scale-up**



- A continuous rotary kiln has been installed and production at pilot scale is being demonstrated
- A cost analysis is underway to estimate the cost of sorbent production

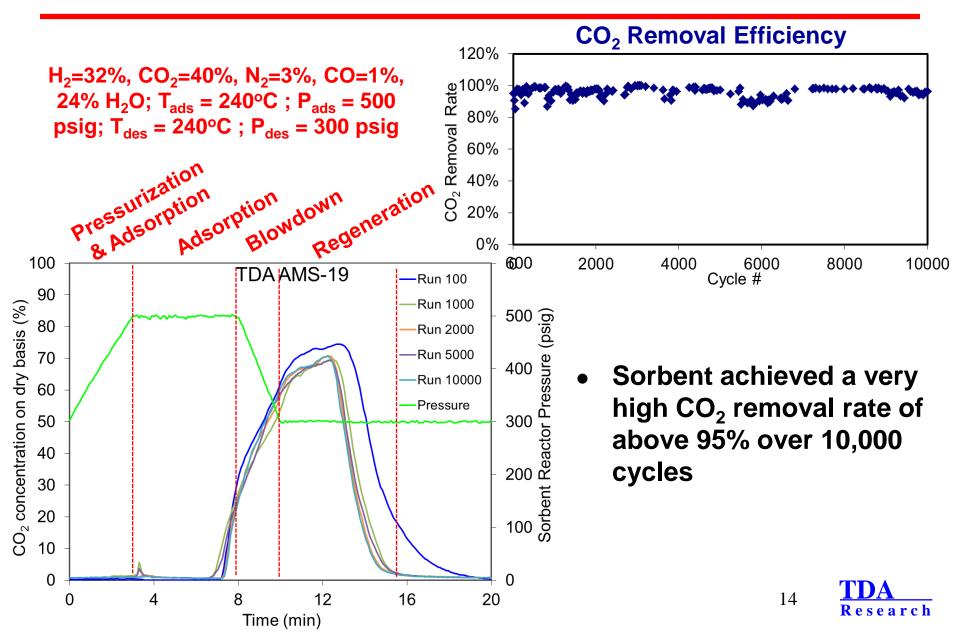
#### **Multiple Cycle Tests**

 $H_2$ =32%,  $CO_2$ =40%,  $N_2$ =3%, CO=1%, $H_2O$ =24%; T= 240°C;  $P_{ads}$ = 500 psig;  $P_{des}$  = 50-300 psig



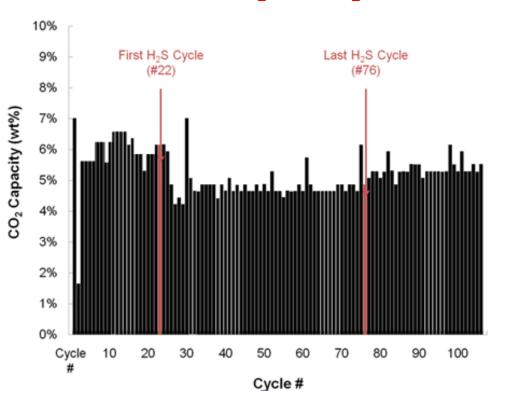
Sorbent maintained its CO<sub>2</sub> capacity (8+%wt.) for more than 10,000 cycles

### CO<sub>2</sub> Removal Efficiency



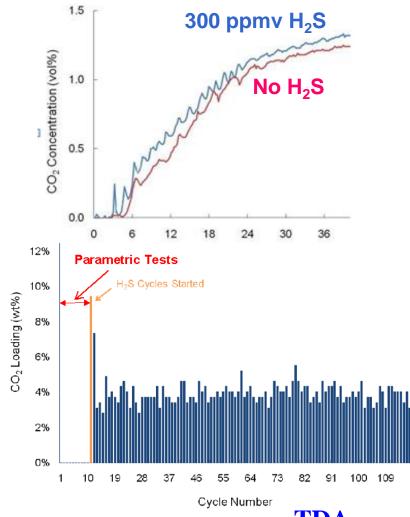
#### Impact of Sulfur

T= 240°C, P= 500 psig, 10 ppmv H<sub>2</sub>S, 44% CO<sub>2</sub>, 20% H<sub>2</sub>, 36% H<sub>2</sub>O; Purge Gas: 50% H<sub>2</sub>, 50% H<sub>2</sub>O



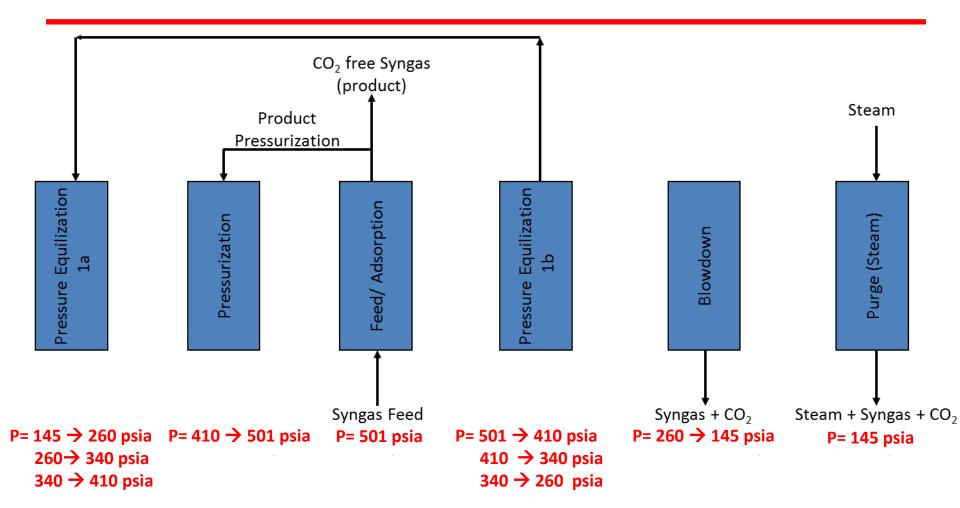
 Presence of H<sub>2</sub>S did not have a significant impact on sorbent performance

300 ppmv H<sub>2</sub>S, T= 240°C, P= 500 psig



15

### **PSA Process Design**



3 pressure equalizations using 8 beds to minimize syngas recycle

### **PSA Cycle Sequence**

#### • PSA Cycle Sequence with 8-beds

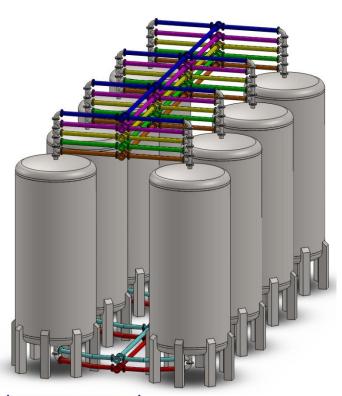
	Sta	ge 1	Sta	ge 2	Stage 3		Stage 4		Stage 5		Stage 6		Stage 7		Stage 8	
Time (min)	2	2	1	1	2	2	1	1	i	2	1	1	2	2	1	1
Bed 1	ADS		EQ1	HOLD	EC	<b>)</b> 2	EQ3	BD	PUI	RGE	EQ4	HOLD	E	<b>Q</b> 5	EQ6	PRESS
Bed 2	EQ6 PRESS ADS		OS	EQ1	HOLD	EQ2		EQ3	BD	PURGE		EQ4 HOLD		EQ5		
Bed 3	EQ5 EQ6 PRESS ADS		)S	EQ1 HOLD EQ2		EQ3	BD	PURGE		EQ4 HOLD						
Bed 4	EQ4	HOLD	EC	<b>Q</b> 5	EQ6	PRESS	ΑI	DS	EQ1	HOLD	E	Q2	EQ3	BD	PU	RGE
Bed 5	PUI	RGE	EQ4	HOLD	EC	<b>Ղ</b> 5	EQ6	PRESS	Al	DS	EQ1	HOLD	E	Q2	EQ3	BD
Bed 6	EQ3	BD	PUI	RGE	EQ4	HOLD	EC	<b>Q</b> 5	EQ6	PRESS	А	DS	EQ1	HOLD	E	Q2
Bed 7	EC	Q2	EQ3	BD	PUF	RGE	EQ4	HOLD	E	<b>Q</b> 5	EQ6	PRESS	Al	DS	EQ1	HOLD
Bed 8	EQ1	HOLD	EC	Q2	EQ3	BD	PUI	RGE	EQ4 HOLD		EQ5		EQ6	PRESS	ADS	

#### 8- bed PSA Cycle Steps:

Step 1	Adsorption at 501 psia (ADS)	Step 6	Steam Purge at 145.1 psia (PURGE)
Step 2	Pressure Equalization to 420 psia (EQ1)	Step 7	Pressure Equalization to 250 psia (EQ4)
Step 3	Pressure Equalization to 340 psia (EQ2)	Step 8	Pressure Equalization to 330 psia (EQ5)
Step 4	Pressure Equalization to 260 psia (EQ3)	Step 9	Pressure Equalization to 410 psia (EQ6)
Step 5	Blowdown to 145.1 psia (BD)	Step 10	Product Pressurization to 501 psia (PRESS)

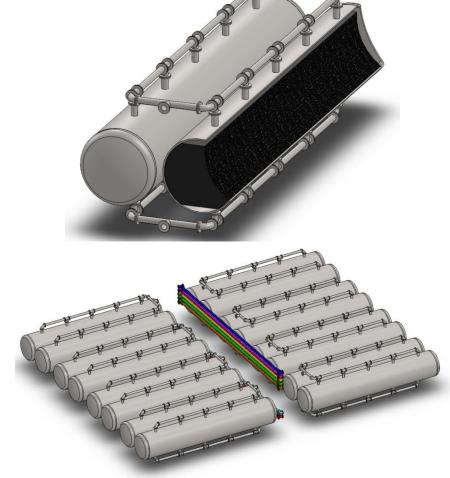
### **Reactor Configurations**

#### High L/D

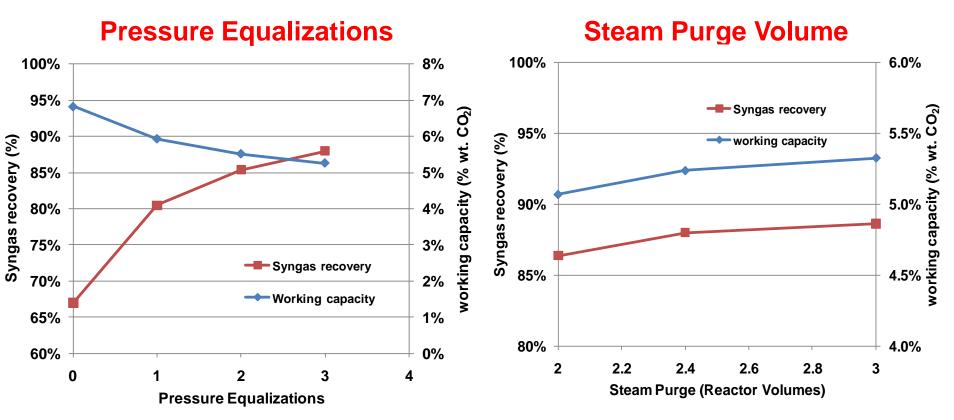


Steam Purge
High Press EQ
Med Press EQ
Low Press EQ
Clean Gas Out
Blowdown/Purge Out
Syngas In





#### **Optimization of Process Parameters**



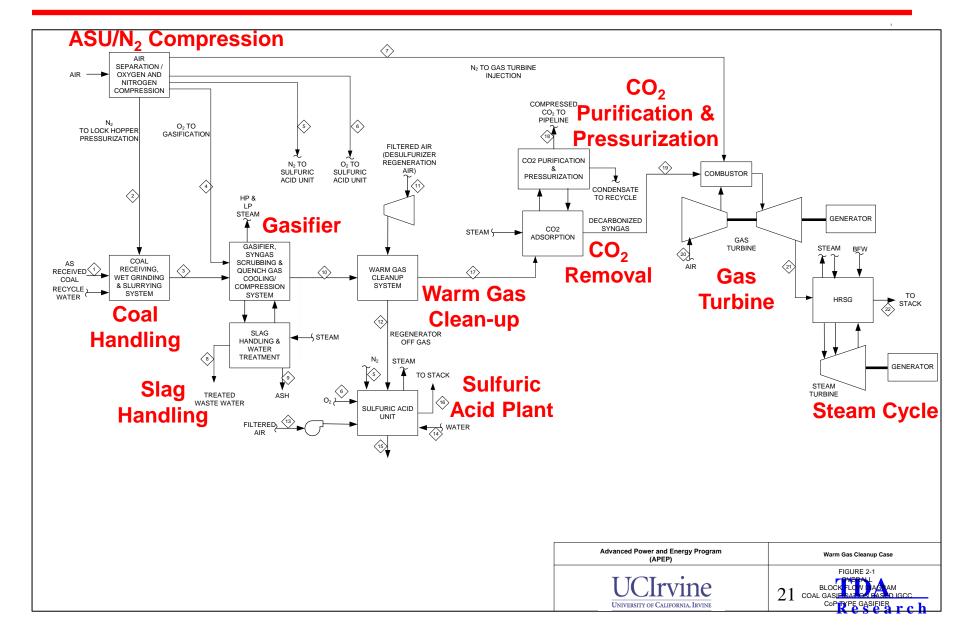
- Three pressure equalization steps are considered to increase synthesis gas recovery
  - Ensures maximum amount of syngas is used as a fuel to gas turbine
- Steam purge volume is being optimized



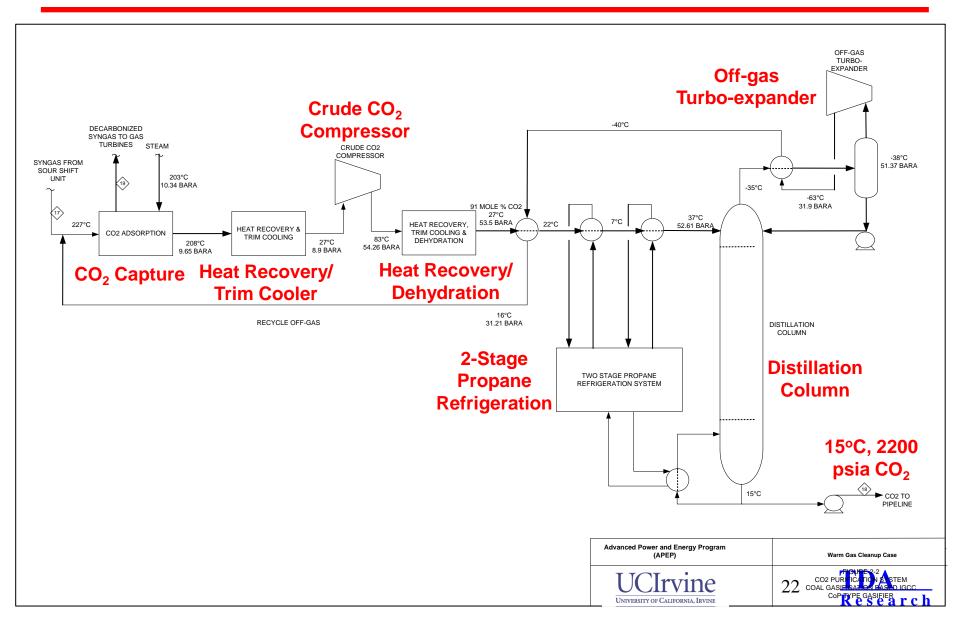
### **System Analysis**

- UCI carries out a process simulation using AspenPlus<sup>™</sup> and evaluate the cost CO<sub>2</sub> capture
- The analysis includes three simulations:
  - E-Gas<sup>™</sup> based IGCC plant with Selexol-based CO<sub>2</sub> capture
    - Calibration Case
    - Compare/validate model results with prior DOE/NETL analysis
  - E-Gas<sup>™</sup> based IGCC plant with Selexol 90% CO<sub>2</sub> capture
  - E-Gas<sup>™</sup> based IGCC plant with TDA's CO<sub>2</sub> capture system
- Same assumptions and cost guidelines will be adopted
  - Consistent design requirements
  - Up-to-date performance and capital cost estimates

### **System Modeling**



### CO<sub>2</sub> Purification & Compression



### **UCI System Analysis Results**

	IGCC-Selexol Calibration Case	IGCC-Selexol 90% Capture	IGCC-TDA-WGC 90% Capture	
CO <sub>2</sub> Capture, %	88.2	90	90	
Gross Power Generated, kWe	696,770	691,624	691,460	
Gas Turbine Power	464,336	461,986	459,990	
Steam Turbine Power	232,434	229,638	231,470	
Auxiliary Load, kWe	171,998	175,498	151,082	
Net Power, kWe	524,772	516,126	540,378	
Net Plant Efficiency, % HHV	32.1	31.6	33.1	

- The IGCC plant with TDA's CO<sub>2</sub> capture technology system achieves higher efficiency than IGCC with Selexol
- Case studies exploring different design configurations on PSA operation, CO<sub>2</sub> purification system

<b>Case Studies</b>	Plant Eff., % HHV
Case 1	32.9
Case 2	32.6
Case 3	32.5
Case 4	32.8
Case 5	32.0

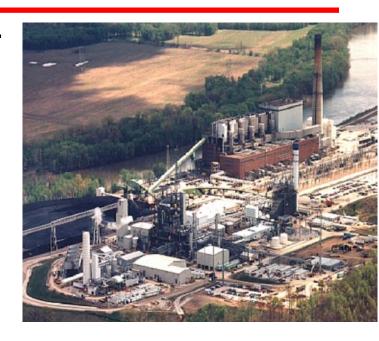
#### **Slipstream Demonstrations**

 Two 3-week test campaigns for proof-ofconcept demonstrations

#### Wabash River IGCC Plant, Terre Haute, IN

- Largest single-train gasifier with 262 MW power output
- Oxy-blown E-Gas<sup>TM</sup> Gasifier
- Operates on petcoke



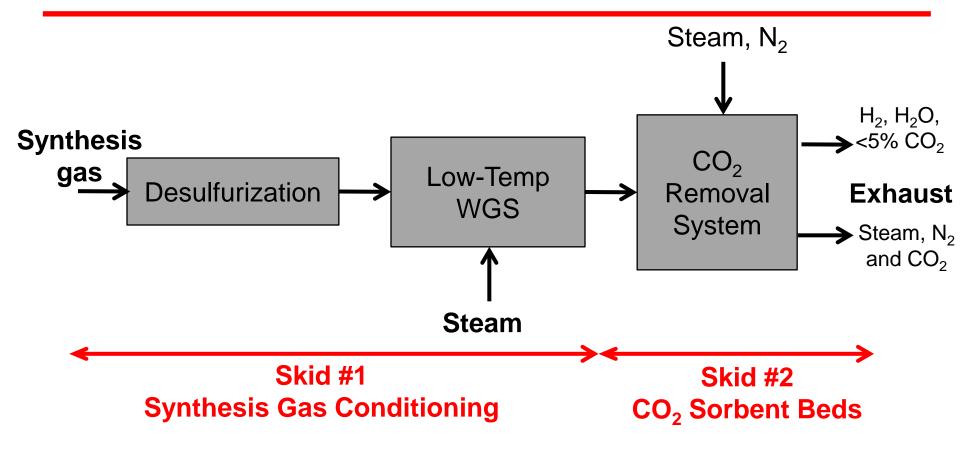


#### National Carbon Capture Center, Wilsonville, AL

- Demonstration starts at October 10, 2011
- Pilot-scale gasifier
- Air-blown transport gasifier (based on KBR's gasification technology)
- Operates on coals and lignites



#### **Slipstream Test Skids**



- Skid #1 Synthesis gas pre-treatment skid
- Skid #2 CO<sub>2</sub> removal skid
- Skid #3 Gas analysis skid

### **System Pictures – Before Insulation**





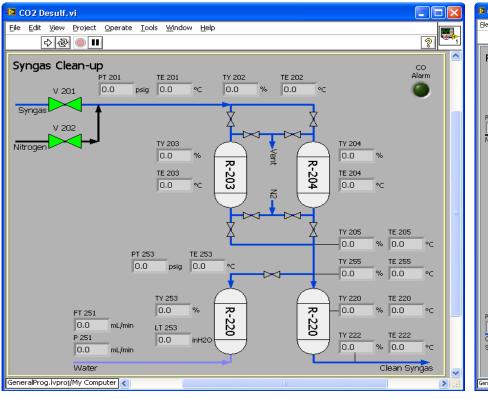
Skid #1

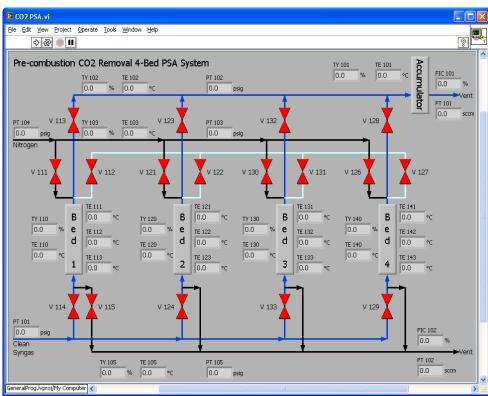
Skid #2

#### **Control System**

Skid #1

#### Skid #2





- System automation is complete
- Ready to move forward with slipstream demonstrations!



#### **Acknowledgments**

- DOE Project Manager
  - Dr. Arun Bose
- TDA Research, Inc.
  - Dr. Steve Dietz, Lauren Brickner, Amanda Parker, Matt Schaefer, Kerry Libberton
- UCI
  - Dr. Ashok Rao
- CoP
  - Dr. Albert Tsang
  - Casey Morriss
- Southern Company
  - Frank Morton
  - Tony Wu
- MWV
  - Paula Walmett